## CONVERGENT APPROACH TO INTERMEDIATES USED FOR THE SYNTHESIS OF THE ENEDIYNE STRUCTURE OF THE ESPERAMICINS AND CALICHEAMICINS

J.F. Kadow\*, M.G. Saulnier, M.M. Tun, D.R. Langley, and D.M. Vyas

Bristol-Myers Company, Pharmaceutical Research and Development Division, 5-Research Parkway, P.O. Box 5100, Wallingford, Connecticut, 06492-7660

Abstract: The diynene lithium acetylide 5 adds to 2-siloxy-2-cyclohexenone 6 followed by silyl migration to produce ketone adduct 7, an intermediate towards a simple Esperamicin core

The disclosures of the structures of the Esperamicins<sup>1</sup> and Calicheamicins<sup>1</sup>, representatives of a new class of extremely potent antitumor antibiotics, have resulted in a flurry of synthetic activity.<sup>2</sup> Efforts have been mainly aimed at the synthesis of bicyclo (7.3.1) tridecane aglycones or simpler monocyclic cores which contain the diynene functionality thought to be responsible for the DNA damaging properties of these cytotoxic compounds. Magnus has published a conceptionally elegant synthesis of the stable bicyclo (7.3.1) tridecane diynene 2.<sup>2b</sup> Recently an independent approach to 2 has appeared.<sup>2c</sup> These communications represent pioneering efforts in the field and prompt us to report a convergent and more efficient synthesis of an advanced intermediate, common to both routes, which should prove useful for future studies.

Both of the published routes to 2 (as well as a synthesis of an alternate core by Kende)<sup>2e</sup> involve a stepwise linear elaboration of a dignene system onto a functionalized cyclohexenone. Herein we report the addition of preconstructed dignene acetylide 5, containing a suitably functionalized propargylic carbon, to 2-tertbutyldimethyl silyloxy-2-cyclohexenone (6) which quickly generates the advanced protected intermediate 7 in a single step. Interestingly, Danishefsky has synthesized a highly functionalized core based on the dilithio anion of the parent 1,5-hexadiyne-3-ene.<sup>2a</sup>

The diynene 4 is assembled by standard Castro-Stephans palladium coupling<sup>3</sup> from the known<sup>2e</sup> <u>cis</u>-chlorc eneyne 3 on decagram scale (Scheme 1). The trimethylsilyl acetylene protecting group was removed <u>in situ</u> prior to anion generation due to the presumed volatility of deprotected 4. The anion 5 is prepared using a procedure similar to that developed by Danishefsky for generation of the parent dilithiodiynene<sup>2d</sup>. Addition of 1.1 eq of 6 to anion 5 in THF at  $-78^{\circ}$ C followed by slow warming to 25°C cleanly generates ketone 7 in 52% yield along with recovered ketone starting material after purification by flash chromatography. The acetylide addition initially provides an intermediate silyl enol ether acetylene adduct which can be isolated at  $-15^{\circ}$ C. Further warming to 25°C allows for complete silyl migration to give the desired product 7.<sup>4</sup> Ketone 6 is prepared in essentially quantitative yield on a 30g scale by reacting 1,2-cyclohexanedione with TBDMSOTf/Et<sub>2</sub>N in THF at -78°C to 25°C followed by filtration through silica gel.

The efficiency of the addition reaction can be increased by utilization of a less volatile diynene, propargyl silyl ether 4b, which was prepared from 4a and purified on a

multigram scale. Lithiation of 4b  $[1.2eq.(TMS)_2NLi, THF, -78°C/15min]$  followed by addition of 6 and warming to 25°C, produced 7a in 87% yield.<sup>5</sup> We have successfully converted our synthetic material 7 into 2 using either of the literature routes<sup>2a,b</sup> and are now in the process of using substrates containing different propargyl substituents for the Magnus cyclization reaction.



a) 3%  $Pd(PPh_3)_4$ , 5% CuI, 2eq. n-BuNH<sub>2</sub>, PhH, 25°, 4h, 71%. b) 1) LiOH, 9:1 THF/H<sub>2</sub>O, 25°, 2h. 2) Pentane extract then Na.SO<sub>4</sub>. 3) 1,10 Phenanthroline, THF, 0° then BuLi until dark. 4) 1.1eq LiN(TMS)<sub>2</sub>, -78° c)  $\mathbf{6}$ , -78° to 25°, then H<sub>2</sub>O, 52% for b and c after chromatography.

Acknowledgement: The authors thank Professor Philip Magnus for communicating his group's research results in this area prior to publication.

## REFERENCES AND NOTES

1. See references contained in reference 2c.

2. a) S.J.Danishefsky, N.B.Mantlo and D.S.Yamashita, J. Am. Chem. Soc., 1988, 110, 6890. b) P. Magnus, R.T.Lewis, and J.C.Huffman, Ibid, 1988, 110, 6921. c) K. Tomioka, H. Fujita, and K.Koga, Tetrahedron Letters, 1989, 30, 851. d) S.J.Danishefsky, D.S.Yamashita, and N.B.Mantlo, Ibid, 1988, 29, 4681. e) A.S.Kende and C.A.Smith, Ibid, 1988, 29, 4217. f) P. Magnus and P.A.Carter, J. Am. Chem. Soc., 1988, 110, 1626. g) K.C. Nicolau, G.Zuccarello, Y. Ogawa, E.J.Scherger, and T. Kumazawa, Ibid, 1988, 110, 4866. h) K.C. Nicolaou, Y.Ogawa, G.Zuccarello, and H.Kataoka, Ibid, 1988, 110, 7247. i) S.L. Schreiber and L.L. Kiesling, Tetrahedron Letters, 1989, 30, 433.

3. a) R.D. Stephans and C.E.Castro, J. Org. Chem., 1963, 28, 3313. b) D.Guillerm and G.Linstrumelle, Tetrahedron Letters, 1986, 27, 5857. c) A.S. Kende and C.A.Smith, J. Org. Chem., 1988, 53, 2655.

4. The yield of 7 from 1,2-cyclohexanedione via our route is 50%, while the reported yields for references 2b and 2c are 29% and 31% respectively.

5. New compounds 4, 4a, 4b, 6, and 7 have been fully characterized spectroscopically and by elemental analysis.

(Received in USA 3 May 1989)